Microwave-Assisted [6+4]-Cycloaddition of Fulvenes and α-Pyrones to Azulene-Indoles: Facile Syntheses of Novel Antineoplastic Agents

Bor-Cherng Hong,* Yea-Fen Jiang and Ellappan Sampath Kumar

Department of Chemistry, National Chung Cheng University, Chia-Yi, 621, Taiwan, ROC

Received 12 April 2001; accepted 8 May 2001

Abstract—A microwave-enhanced [6+4]-cycloaddition reaction between 6-aminofulvene and pyrones followed by CO₂ extrusion provides azulene–indoles which display interesting antineoplastic activity. © 2001 Elsevier Science Ltd. All rights reserved.

The widespread occurrence and interesting biological activities of indoles and azulenes in nature make them important targets for synthesis. In previous studies, we described a novel synthesis of azulene–indoles via a unique cycloaddition reaction of fulvene–ketene acetals and α -pyrones (Scheme 1). The formation of the azulene–indoles is envisaged to occur via a [6+4]-cycloaddition of fulvene with α -pyrone followed by cheleotropic extrusion of CO_2 . This reaction provides a rapid construction of functionalized polycyclic azulene–indole systems with potential application to solid-phase synthesis and combinatorial libraries of these skeletons. Unfortunately, our initial results indicated that, with conventional heating, reaction times were excessive (>4 days) and product yields were low (10–25%).

Microwave-assisted organic synthesis has received much attention in recent years.^{3,4} It has been demonstrated that some organic reactions proceed faster and with higher efficiency under microwave irradiation as opposed to conventional heating. The use of open focused microwave ovens⁵ has added further to the evolution of this technique for accelerating thermal organic reactions. The application of microwave irradiation to the Diels–Alder reaction is also well-documented.⁶ However, to the best of our knowledge, microwave-assisted higher-order cycloaddition reactions⁷ (e.g., [6+4]-cycloaddition)⁸ have not yet been explored.

In an initial experiment, a DMSO solution of 6-dimethylaminofulvene (1) and α -pyrone (2)¹⁰ was irradiated by using a focused microwave oven (10% power, 130 °C, 1 h) to provide 32% of the azulene-indole 3.11 Experiments were carried out to test the effect of solvents and conditions on the yield of this process (Table 1). In general, DMF appears to be the best solvent. The highest yield was observed for the reaction of 1 (1 M in DMF) and 2 with 10% microwave power for 60 min at 130 °C (entry 12, 65% isolated product). Reactions performed at higher microwave power and for longer irradiation times gave lower yields of the azulene-indoles along with decomposition products (entries 3, 5, and 7). In addition, lower yields were observed when reactant concentrations were higher or lower than 1 M (entries 8, 10, and 11). It is noteworthy that reactions performed

Scheme 1. Azulene–indole forming fulvene–ketene α -pyrones [6+4]-cycloaddition.

In conjunction with our continuing investigations in the area of fulvene chemistry, 9 we explored the use of microwave irradiation in [6+4]-cycloaddition reactions between 6-dimethylaminofulvene and α -pyrones. We describe herein the application of this methodology to the synthesis of novel antineoplastic azulene—indole systems. Preliminary structure—activity relationships against a variety of human cancer cell lines are also reported.

^{*}Corresponding author. Tel.: +886-5-242-8174; fax: +886-5-272-1040; e-mail: chebch@ccunix.ccu.edu.tw

Table 1. Microwave-assisted [6+4]-cycloaddition of 6-aminofulvene 1a and pyrone 2a to azulene-indole 3

Entry	Fulvene:Pyrone (molar ratio)	Solvent	Power (%)	Concn (M)	Time (min)	Temp. (°C)	Yielda (%)
1	1:1	DMSO	0ь	0.5	7200	130	8
2	1:1	DMSO	10	0.5	60	130	32
3	1:1	DMSO	10	0.5	60	160	27
4	1:1	DMF	$0_{\rm p}$	0.5	7200	130	22
5	1:1	DMF	10	0.5	180	130	48
6	1:1	DMF	10	0.5	60	130	50
7	1:1	DMF	30	0.5	60	130	38
8	1:1	DMF	10	0.25	60	130	30
9	1:1.5	DMF	10	1	60	130	57
10	1:1.5	DMF	10	2	60	130	45
11	1:1.5	DMF	10	4	60	130	30
12	5:1	DMF	10	1	60	130	65°

^aIsolated yields based on starting fulvene.

3. R₂ = H, R₃ = CH₃, R₄ = H 1a. R₁ = NMe₂; R₂ = H 2a. R₃ = CH₃; R₄ = H 1b. R₁ = R₂ = -OCH₂CH₂O-**2b.** $R_3 = Et$; $R_4 = H$ 4. R₂ = H, R₃ = Et, R₄ = H 5. R₂ = H, R₃ = Ph, R₄ = H 1c. R₁ = NMe₂; R₂ = CH₃ **2c.** R₃ = Ph; R₄ = H 6. R₂ = CH₃, R₃ = CH₃, R₄ = H **2d.** R₃ = CH₃; R₄ = OAc **1d.** $R_1 = NMe_2$; $R_2 = NMe_2$ 7. R_2 = -OCH $_2$ CH $_2$ OH, R_3 = CH $_3$, R_4 = H 2e. R₃ = CH₃; R₄ = Br 8. R₂ = -OCH₂CH₂OAc, R₃ = CH₃, R₄ = H $R_2 = -OCH_2CH_2OC(=O)N(CH_2CH_2CI)_2,$ $R_3 = CH_3, R_4 = H$ a. DMF, Microwave **b**. Ac₂O, Et₃N, CH₂Cl₂, DMAP; 90% 10. $R_2 = -OCH_2CH_2OH$, $R_3 = Et$, $R_4 = H$ 11. R₂ = -NMe₂, R₃ = CH₃, R₄ = H c. CIC(=O)N(CH2CH2CI)2, DBU, CH₂Cl₂, DMAP; 90%

d. cat. LiOH, MeOH; 90% 12. R₂ = H, R₃ = CH₃, R₄ = OAc 13. R₂ = H, R₃ = CH₃, R₄ = OH 14. R₂ = H, R₃ = CH₃, R₄ = Br

Scheme 2. Syntheses of various azulene-indoles.

Table 2. [6+4]-Cycloaddition of 1 and 2 to azulene–indole 3

Entry	Fulvene + Pyrone	Product no.	Yield ^{a,b} (%) No MW	Yield ^{b,c} (%) with MW	
1	1a + 2a	3	26	65	
2	1a + 2b	4	18	46	
3	1a + 2c	5	12	30	
4	1c + 2a	6	23	57	
5	1b + 2a	7	19	52	
6	1b + 2b	10	18	48	
7	1d + 2a	11	14	37	
8	1a + 2d	12	26	62	
9	1a + 2e	14	21	53	

^aConventional oil bath heating without microwave irradiation for 7200 min

Table 3. In vitro inhibition of cancer cell lines by azulene–indoles^{a,b}

Compound	NSC	Average IC_{50} (μM) ^c	Selected IC_{50} $(\mu M)^{d,e}$
7	695154	2.51	EKVX (1.00), HS578T (0.93), NCI-H322M (17.0), A498 (13.8)
3	710184	2.40	U251 (1.48), UO-31 (1.62), A498 (11.7), CAK1-1 (6.6)
12	714067	2.19	NCI-H226 (1.38), IGROV1 (1.58), SR (4.57), K-562 (3.80)
14	716773	5.37	T-47D (1.70), EKVX (1.78), UACC-257 (24.0), A498 (21.9)
11	718564	3.98	HOP-92 (0.56), PC-3 (1.41), A498 (20.9), CAK1-1 (17.4)
8	710185	2.75	K-562 (0.51), HOP-92 (1.38), NCI-H322M (14.5), NCI-H226 (11.2)

^aResults of NCI's in vitro disease-oriented tumor cell line screen assays.

by using conventional oil bath heating require long reaction times (5 days) and give low yields of the final product along with recovered starting materials and decomposition products (entries 1 and 4). As shown in Scheme 2 and Table 2, a series of azulene–indoles were synthesized with the same reaction conditions.¹²

The azulene–indoles, prepared in this study, were evaluated according to standard NCI protocols¹³ for their in vitro activity against 60 human cancer cell lines derived from 10 clinically isolated cancer subtypes (melanoma, leukemia, lung, colon, renal, ovarian, CNS, brain, breast and prostate). As shown in Table 3, these

^bConventional oil bath heating without microwave irradiation.

^cIsolated yield based on starting pyrone.

^bIsolated yields based on starting fulvene.

^cMicrowave irradiation for 60 min.

 $[^]b\mathrm{IC}_{50}\!\!:$ concentration that inhibits 50% cell growth.

^cAverage IC₅₀ against 60 human cell lines.

^dSelected highest and lowest activities against various cell lines.

^eThe cancer cell lines are defined as follows: EKVX, HOP-92, NCI-H226 and NCI-H322M are non-small cell lung; HS578T and T-47D are breast; A498, UO-31 and CAK1-1 are renal; U251 is CNS, SR and K-562 are leukemia; UACC-257 is melanoma; IGROV1 is ovarian; PC-3 is prostate.

compounds display good activity against most of the cell lines tested, with average IC $_{50}$ values ranging between 2.2 and 5.4 μM . Also, some of the azulene–indoles show selective growth inhibitory activity. For example, compound 8 is a potent and selective growth inhibitor of the K-562 leukemia cell line (IC $_{50} = 0.51 \, \mu M$). Results from the COMPARE program of the NCI screening database seem to indicate that these azulene–indoles represent a class of antitumor agents with a novel mechanism of action. 14

In summary, a facile and efficient microwave-assisted [6+4]-cycloaddition method for the synthesis of antineoplastic azulene-indoles has been developed. The potential biological activity of other members of this family as well as the mechanism¹⁵ of action of these compounds are under active investigation.

Acknowledgements

The authors are grateful to the staff of the National Cancer Institute, USA, and the staff of the National Health Research Institute, ROC for the anticancer testing studies. Thanks are also given to Dr. Sepehr Sarshar and Professor Ta-Jun Lu for valuable discussions. Mass spectral data were provided by the National Science Council Spectroscopic Service Center. Financial support from National Science Council, National Health Research Institute and National Chung Cheng University are gratefully acknowledged.

References and Notes

- 1. (a) For a review of recent developments in indole synthetic methodologies, see: Gribble, G. W. J. Chem. Soc., Perkin Trans. 1 2000, 1045. (b) For a recent example of the synthesis of azulene derivatives, see: Ito, S.; Inabe, H.; Okujima, T.; Morita, N.; Watanabe, M.; Imafuku, K. Tetrahedron Lett. 2000, 41, 8343. (c) For a review on the synthesis of azulenes, see: Fleming, P. R.; Sharpless, K. B. Chemtracts: Org. Chem. 1988, 2, 205.
- 2. Hong, B.-C.; Sun, S.-S. Chem. Commun. 1996, 937.
- 3. (a) For reviews, see: Varma, R. S. *Green Chem.* **1999**, *1*, 43. (b) Bose, A. K.; Banik, B. K.; Lavinskaia, N.; Jayaraman, M.; Manhas, M. S. *Chemtech* **1997**, 27, 18. (c) Caddick, S. *Tetrahedron* **1995**, 51, 10403. (d) Strauss, C. R.; Trainor, R. W. *Aust. J. Chem.* **1995**, 48, 1665. (e) Galema, S. A. *Chem. Soc. Rev.* **1997**, 26, 233.
- 4. (a) For selected manuscripts published in 2000 regarding microwave-assisted organic reactions (MAOR), see: Kabza, K. G.; Chapados, B. R.; Gestwicki, J. E.; McGrath, J. L. J. Org. Chem. 2000, 65, 1210. (b) Gibbons, K. E.; Jones, M. O.; Blundell, S. J.; Mihut, A. I.; Gameson, I.; Edwards, P. P.; Miyazaki, Y.; Hyatt, N. C.; Porch, A. Chem. Commun 2000, 159. (c) Gadhwal, S.; Baruah, M.; Prajapati, D.; Sandhu, J. S. Synlett 2000, 341. (d) Ganschow, M.; Wark, M.; Woehrle, D.; Schulz-Ekloff, G. Angew. Chem., Int. Ed. 2000, 39, 161. Williams, L. Chem. Commun. 2000, 435. (e) Langa, F.; Cruz, P.; Hoz, A.; Espildora, E.; Cossio, F. P.; Lecea, B. J. Org. Chem. 2000, 65, 2499. (f) Olsson, R.; Hansen, H. C.; Andersson, C.-M. Tetrahedron Lett. 2000, 41, 7947. (g) Sauvagnat, B.; Lamaty, F.; Lazaro, R.; Martinez, J. Tetrahedron Lett. 2000,

- 41, 6371. (h) Soujri, M.; Guillaumet, G.; Besson, T.; Aziane, D.; Aadil, M.; Essassi, E. M.; Akssira, M. Tetrahedron Lett. 2000, 41, 5857. (i) Lee, J. C.; Song, I.-G. Tetrahedron Lett. 2000, 41, 5891. (j) Kabalka, G. W.; Wang, L.; Namboodiri, V.; Pagni, R. M. Tetrahedron Lett. 2000, 41, 5151. (k) Usyatinsky, A. Y.; Khmelnitsky, Y. L. Tetrahedron Lett. 2000, 41, 5031. (1) Sharma, U.; Ahmed, S.; Boruah, R. C. Tetrahedron Lett. 2000, 41, 3493. (m) Moghaddam, F. M.; Dakamin, M. G. Tetrahedron Lett. 2000, 41, 3479. Kabza, K. G.; Chapados, B. R.; Gestwicki, J. E.; McGrath, J. L. J. Org. Chem. 2000, 65, 1210. (n) Seijas, J. A.; Vazquez-Tato, M. P.; Martinez, M. M. Tetrahedron Lett. 2000, 41, 2215. (o) Marquez, H.; Perez, E. R.; Plutin, A. M.; Morales, M. Tetrahedron Lett. 2000, 41, 1753. (p) Bentiss, F.; Lagrenee, M.; Barbry, D. Tetrahedron Lett. 2000, 41, 1539. (q) Besson, T.; Guillard, J.; Rees, C. W. Tetrahedron Lett. 2000, 41, 1027. (r) Kiddle, J. J. Tetrahedron Lett. 2000, 41, 1339. (s) Ranu, B. C.; Hajra, A.; Jana, U. Tetrahedron Lett. 2000, 41, 531.
- 5. (a) Commarmot, R.; Didenot, R.; Gardais, J. F. French patent 2,560,529, 1985, Rhône-Poulenc/Prolabo. (b) Jacquault, P. (Prolabo company) European Patent 545995 AI (21-12-92), 1992.
- 6. (a) For examples, see: Avalos, M.; Babiano, R.; Cintas, P.; Clemente, F. R.; Jimenez, J. L.; Palacios, J. C.; Sanchez, J. B. J. Org. Chem. 1999, 64, 6297. (b) Sasaki, S. I.; Ishibashi, N.; Kuwamura, T.; Sano, H.; Matoba, M.; Nisikawa, T.; Maeda, M. Bioorg. Med. Chem. Lett. 1998, 8, 2983. (c) Diaz-Ortiz, A.; Carrillo, J. R.; Gomez-Escalonilla, M. J.; De la Hoz, A.; Moreno, A.; Prieto, P. Synlett 1998, 1069. (d) Sridhar, M.; Krishna, K. L.; Srinivas, K.; Rao, J. M. Tetrahedron Lett. 1998, 39, 6529. (e) Fernandez-Paniagua, U. M.; Illescas, B.; Martin, N.; Seoane, C.; de la Cruz, P.; de la Hoz, A.; Langa, F. J. Org. Chem. 1997, 62, 3705.
- 7. Cycloadditions play a central role in organic chemistry and recent advances in higher-order cycloadditions, such as [5+2], [6+2] and [6+4], have further expanded their use in annulation strategies. For a recent special report on this topic, see: Zurer, P. S. Chem. Eng. News **2001**, 79, 27.
- 8. For a recent review of [6+4] cycloadditions and their applications in total synthesis, see: Gallow, F.; Paquette, L. A. *Chemtracts* **2000**, *13*, 223.
- 9. (a) For previous papers in this series, see: Hong, B.-C.; Shen, I.-C.; Liao, J.-H. *Tetrahedron Lett.* **2001**, 42, 935. (b) Hong, B.-C.; Chen, Z.-Y.; Chen, W.-H. *Org. Lett.* **2000**, 2, 2647. (c) Hong, B.-C.; Sun, H.-I.; Chen, Z.-Y. *Chem. Commun.* **1999**, 2125. (d) Hong, B.-C.; Chen, Z.-Y.; Kumar, E. S. *J. Chem. Soc., Perkin Trans. 1* **1999**, 1135. (e) Hong, B.-C.; Hong, J.-H. *Tetrahedron Lett.* **1997**, 38, 255. (f) Hong, B.-C.; Sun, S.-S.; Tsai, Y.-C. *J. Org. Chem.* **1997**, 62, 7717.
- 10. (a) For the preparation of α-pyrone, see: Moody, C. J.; Shah, P. J. Chem. Soc., Perkin Trans. 1 1988, 1407. (b) Moody, C. J.; Shah, P.; Knowles, P. J. Chem. Soc., Perkin Trans. 1 1988, 3249. (c) Medio-Simon, M.; Erfanian-Abdoust, H.; Pindur, U. Liebigs Ann. Chem 1991, 601. (d) Jackson, P. M.; Moody, C. J.; Mortimer, R. J. J. Chem. Soc., Perkin Trans. 1 1991, 2941. (e) Pindur, U.; Erfanian-Abdoust, H. Liebigs Ann. Chem. 1990, 771.
- 11. Focused microwave irradiation was carried out at atmospheric pressure with a Synthewave S402 Prolabo microwave reactor (300 W, monomode system, 10 mL reactors). The apparatus has a quartz reactor, visual control, PC controlled 300 W irradiation, infrared temperature measurement with continuous feedback control.
- 12. All new compounds were characterized by full spectroscopic data (¹H, ¹³C, NMR, IR, MS, and HR-MS). **Typical procedure for synthesis of azulene-indole 3**: A mixture of 6-dimethylaminofulvene (**1**, 604 mg, 5 mmol), 1-methylpyrano[3,4-*b*]indol-3(9*H*)-one (**2**, 200 mg, 1 mmol) and DMF (5 mL) were placed in a 10 mL quartz vial and subjected to

programmed microwave irradiation at 30 W for 60 min. After a period of 2-3 min, the temperature reached a plateau of 130 °C where it remained throughout the reaction. After cooling, the solution was concentrated and the residue was subjected to flash column chromatography (10% EtOAc-hexane, $R_f = 0.48$ in 30% EtOAc-hexane) to give the azuleneindole 3 as a dark green solid (150 mg, 65% yield) and recovered fulvene (490 mg). Spectroscopic data for 3: mp 157-159 °C; IR (neat): 3416, 3077, 2937, 2857, 1626, 1507, 1462, 1377, 1302, 1237, 751 cm⁻¹; ${}^{1}H$ NMR ($C_{6}D_{6}$, 400 MHz): δ 8.24 (d, $J=10 \,\mathrm{Hz}$, 1H), 8.01 (d, $J=8 \,\mathrm{Hz}$, 1H), 7.94 (t, J = 3.5 Hz, 1H), 7.78 (d, J = 10 Hz, 1H), 7.54–7.50 (m, 3H), 7.38 (dd, J = 8, 8 Hz, 1H), 7.25–7.10 (m, 1H), 7.07 (dd, J = 8.0, $0.6\,\mathrm{Hz},\,1\mathrm{H}),\,2.41$ (s, 3H); $^{13}\mathrm{C}$ NMR ($\mathrm{C}_6\mathrm{D}_6,\,100\,\mathrm{MHz}$): δ 139.0 (C), 138.5 (C), 137.9 (C), 135.2 (C), 134.4 (CH and C), 129.8 (CH), 128.5 (C), 127.9 (CH), 127.4 (C), 121.6 (CH), 121.1 (CH), 119.0 (CH), 114.5 (CH), 114.3 (CH), 111.7 (CH), 19.0 (CH₃); MS (m/z, relative intensity): 231 (M⁺, 17), 173 (100), 158 (90), 149 (39), 130 (42); exact mass calcd for $C_{17}H_{13}N$ (M⁺): 231.1049; found 231.1048.

13. Assays were performed at the National Cancer Institute (USA). For details of the assay protocols, see: Grever, M. R.; Schepartz, S. A.; Chabner, B. A. Semin. Oncol. 1992, 19, 622.

- (b) Monks, A.; Scudiero, D. A.; Skehan, P.; Shoemaker, R. H.; Paull, K. D.; Vistica, D. T.; Hose, C.; Langley, J.; Cronice, P.; Vaigro-Wolf, M.; Gray-Goodrich, M.; Campbell, H.; Mayo, M. R. J. Natl. Cancer Inst. 1991, 83, 757.
- 14. These compounds lie outside the category of common antitumor agents. COMPARE is a method for analyzing the results of the NCI Human Tumor Cell Line Screens. In general, high correlation of screening data translates into similar mechanism of action. For reference, see: Hodes, L.; Paull, K. D.; Koutsoukos, A.; Rubinstein, L. R. *J. Biopharm. Stat.* **1992**, *2*, 31.
- 15. Cheng, C. C. Unpublished results. Due to the broad antineoplastic activity of these compounds, we carried out some experiments in order to rule out DNA intercalation. No significant unwinding associated with DNA intercalation was observed in an agarose gel study using supercoiled plasmid DNA. In addition, incubation of calf thymus DNA (60 mM per nucleotide) with azulene–indoles (10 mM) resulted in a small change in the melting temperature ($<2^{\circ}$ C). Control experiments with ethidium bromide resulted in a 12–13 °C $T_{\rm m}$ change. The binding constants of azulene–indoles to calf thymus DNA were found to be ca. 102 M⁻¹ (spectral titration at 320 nm in 10 mM phosphate buffer at pH 7).